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which gas particles pass through a tiny opening. The rates of effusion of different gases are directly proportional to the velocities of their particles.

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1. Gases consist of large numbers of tiny particles that are far apart relative to their size. 2. Collisions between gas particles and container walls are elastic collisions. 3. Gas particles are in continuous, rapid, random motion. They therefore possess kinetic energy.

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10. If the temperature of a container of gas remains constant, then the pressure of the gas will increase if (a) the mass of the gas molecules increases; (b) the diffusion of the gas molecules increases; (c) the size of the container is increased; (d) the number of gas molecules in the container is increased. 11.

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1. Gases consist of large numbers of tiny particles that are far apart relative to their size. 2. Collisions are elastic between particles and the container. 3. Particles are always in motion. No attractive forces between them except near condensation point. 4. No forces of attraction or repulsion between particles.

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Chapter 10 - Physical Characteristics of Gases

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• Primary sex characteristics: structures that are present at birth
Female primary sex characteristics - vagina: the tube that leads from

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the outside of a female's body to the opening of the womb - uterus:
the womb in which the baby grows during pregnancy - ovaries: the
female sexual glands Male primary sex ...

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classification of people as male or female based on biological characteristics. ... Chapter 10 - Inequalities of Gender and Age 9 Terms. ekulpins. Sociology age test 16 Terms. samanthaapaigee. Sociology Chapter 10 Vocab 19 Terms. Mitchel_Becher. Sociology Chapters 9 & 10 36 Terms.

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Chapter 10: Introduction to the Lithosphere

Chapter 10 Review Physical Characteristics CHAPTER 10 REVIEW Physical Characteristics of Gases MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Pressure can be represented by the following equation: pressure increase a. For a constant area, as the force increases the pressure will . decrease b.

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